

9.17

$$h = 20 \text{ m}$$

$$v = 3 \text{ m/s}$$

$$c = 130 \text{ J/kgK}$$

$$\Delta T = ?$$



energijski zakon: $\Delta W = A' + Q$ ← *teplota*

↓

$\Delta W_k + \Delta W_p + \Delta W_{pr} + \Delta W_m + \dots$ ← *možnaya energija*

Intuitivno, s katero se krogljica zaleti v tla:

postavke: $\Delta W_p + \Delta W_k = 0$

$$-mgh + \frac{1}{2}mv_0^2 = 0 \rightarrow v_0 = \sqrt{2gh} \sim 20 \text{ m/s}$$

$$\Delta W_k + \Delta W_p + \Delta W_m = 0$$

$$\frac{1}{2}mv^2 - mgh + \Delta W_m = 0$$

$$mc\Delta T = \Delta W_m = mgh - \frac{1}{2}mv^2$$

$$\Delta T = \frac{gh - \frac{1}{2}v^2}{c} = \frac{9,81 \frac{\text{m}}{\text{s}^2} \cdot 20 \text{ m} - \frac{1}{2}(3 \text{ m/s})^2}{130 \text{ J/kgK}}$$

$$= \underline{\underline{1,5 \text{ K} = 1,5^\circ\text{C}}}$$

$$\frac{\text{m}^2/\text{s}^2}{\text{J/kgK}} = \frac{\text{m}^2 \text{ s}^2 \text{ kg K}}{\text{s}^2 \cdot \text{kg m}^2} = \text{K}$$

9.26

- $m = 1 \text{ kg}$
- $V_1 = 1 \text{ m}^3$
- $p_1 = 10^5 \text{ Pa}$
- $V_2 = 2 \text{ m}^3$
- $p_2 = 1,5 \text{ bar}$
- $A_{opr} = 0,1 \text{ MJ}$

$Q_{pr} = ?$

projekto delo
 $\Delta W_m = A + Q \rightarrow$ projekta funkcija
 $Q_{pr} = Q = \Delta W_m - A = \Delta W_m + A_{opr}$
 $A = -A_{opr}$

idealni plin: $\Delta W_m = m c_v \Delta T$

$Q_{pr} = m c_v (T_2 - T_1) + A_{opr}$

$dW_m = dA + dQ = -pdV + dQ$

V konst. $\rightarrow dV = 0$

$dW_m = dQ = m c_v dT$

p konst. \rightarrow

$m c_v dT = dW_m = -pdV + dQ = -pdV + m c_p dT$

$pV = \frac{m}{M} RT$

p konst.: $pdV = \frac{m}{M} R dT$

$m c_v dT = -\frac{m}{M} R dT + m c_p dT$

za idealni plin
 $c_v = -\frac{R}{M} + c_p$

$\frac{c_p}{c_v} = \kappa$

za dvoatomne molekule (npr. O_2, N_2): $\kappa = \frac{7}{5} = 1,4$

$c_v = -\frac{R}{M} + \kappa c_v$

$c_v (\kappa - 1) = \frac{R}{M}$

$c_v = \frac{R}{M(\kappa - 1)}$

$Q_{pr} = m c_v (T_2 - T_1) + A_{opr}$

$Q_{pr} = m \frac{R (T_2 - T_1)}{M(\kappa - 1)} + A_{opr}$

$Q_{pr} = \frac{1}{\kappa - 1} (p_2 V_2 - p_1 V_1) + A_{opr}$

$Q_{pr} = \frac{1}{0,4} (1,5 \cdot 10^5 \text{ Pa} \cdot 2 \text{ m}^3 - 10^5 \text{ Pa} \cdot 1 \text{ m}^3) + 0,1 \text{ MJ} =$

$= \frac{1}{0,4} \cdot 2 \cdot 10^5 \text{ J} + 0,1 \text{ MJ} = 5 \cdot 10^5 \text{ J} + 0,1 \text{ MJ} = \underline{\underline{0,6 \text{ MJ}}}$

$\text{Pa} \cdot \text{m}^3 = \frac{\text{N}}{\text{m}^2} \cdot \text{m}^3 =$
 $= \text{Nm} = \text{J}$

9.29 $m = 1 \text{ kg}$

$T_1 = 20^\circ\text{C}$

$P_1 = 10^5 \text{ Pa}$ ← hitro razpnevo (adiabata)

$P_2 = 8 \times 10^4 \text{ Pa}$ ← stisnemo pri konstantni temperaturi (izoterma)

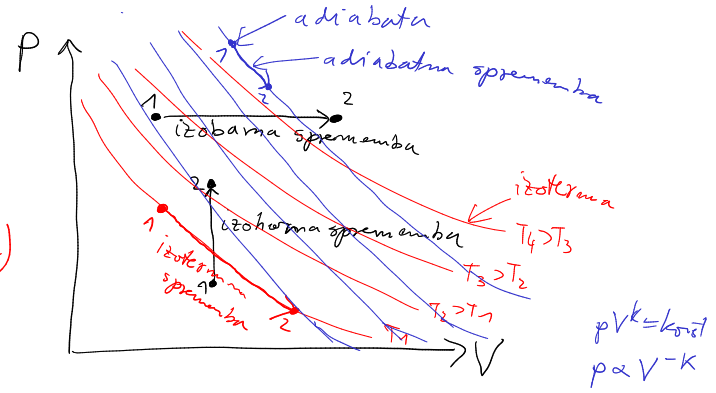
$P_3 = P_1 = 10^5 \text{ Pa}$

$T_2 = T_3$

$\Delta W_m = ?$

$A_{pr} = ?$

$Q_{odd} = ?$



$T = \text{konst.}$ izotermna sprememba

za idealni plin:

$$pV = \frac{m}{M} RT \rightarrow \frac{pV}{T} = \left(\frac{m}{M}\right) R = \text{konst.}$$

$$T = \text{konst.} \rightarrow pV = \text{konst.}$$

$dQ = 0$ adiabatna sprememba

$$dW_m = m c_v dT = -p dV$$

$$c_v = \frac{R}{M(k-1)}$$

$$\frac{mR}{M(k-1)} dT = -p dV$$

$$pV = \frac{m}{M} RT \rightarrow dpV + p dV = \frac{m}{M} R dT$$

$$\frac{dpV + p dV}{k-1} = -p dV / (k-1)$$

$$dpV + p dV = (1-k) p dV$$

$$dpV = -k p dV$$

$$\frac{dp}{p} = -k \frac{dV}{V} \quad / \int$$

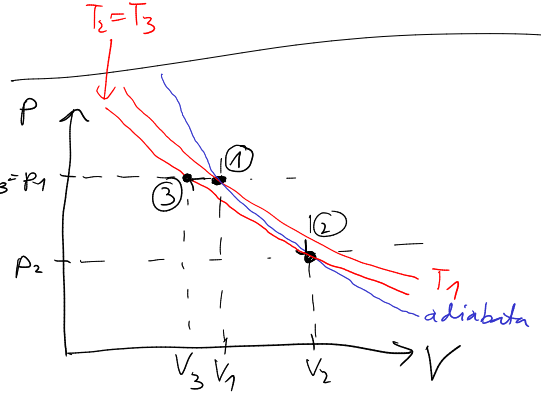
$$\int_{P_1}^{P_2} \frac{dp}{p} = \ln \frac{P_2}{P_1} = -k \int_{V_1}^{V_2} \frac{dV}{V} = -k \ln \frac{V_2}{V_1} / e^{\square}$$

$$\frac{P_2}{P_1} = \left(\frac{V_2}{V_1}\right)^{-k}$$

$$P_1 V_1^k = P_2 V_2^k$$

$$pV^k = \text{konst.}$$

za idealni plin:



$$\Delta W_m = \Delta W_m^{1 \rightarrow 2} + \Delta W_m^{2 \rightarrow 3} = m c_v (T_2 - T_1) + 0 = m \frac{R(T_2 - T_1)}{M(k-1)}$$

adiabata:

$$P_1 V_1^k = P_2 V_2^k$$

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad / \square^k$$

$$\frac{P_1^k V_1^k}{T_1^k} = \frac{P_2^k V_2^k}{T_2^k}$$

$$P_1^{1-k} T_1^k = P_2^{1-k} T_2^k$$

$$T_2 = T_1 \left(\frac{P_1}{P_2}\right)^{\frac{1-k}{k}}$$

$$\Delta W_m = m \frac{R T_1}{M(k-1)} \left(\left(\frac{P_1}{P_2}\right)^{\frac{1-k}{k}} - 1 \right) = \frac{1 \text{ kg} \cdot 8300 \text{ J} \cdot 293 \text{ K}}{8.314 \text{ J/mol} \cdot 0.4 \cdot 29 \text{ g/mol}} \left(\left(\frac{10^5 \text{ Pa}}{8 \times 10^4 \text{ Pa}}\right)^{\frac{1-1.4}{1.4}} - 1 \right) = -13 \text{ kJ}$$

$$Q_{pr} = Q_{pr}^{1 \rightarrow 2} + Q_{pr}^{2 \rightarrow 3} = 0 + Q_{pr}^{2 \rightarrow 3}$$

$$0 = \Delta W_m^{2 \rightarrow 3} = A^{2 \rightarrow 3} + Q^{2 \rightarrow 3} \rightarrow Q^{2 \rightarrow 3} = -A^{2 \rightarrow 3}$$

$$dQ = -dA = p dV$$

$$Q_{pr} = Q_{pr}^{2 \rightarrow 3} = -A_{pr}^{2 \rightarrow 3} = \int_{V_2}^{V_3} p dV = \frac{m}{M} R T_2 \int_{V_2}^{V_3} \frac{dV}{V} = p_2 V_2 \ln \frac{V_3}{V_2} = p_2 V_2 \ln \frac{P_2}{P_3} =$$

zrak: O_2 20% $M_{O_2} = 32 \text{ g/mol}$
 N_2 80% $M_{N_2} = 28 \text{ g/mol}$
 $\rightarrow \bar{M} = 20\% M_{O_2} + 80\% M_{N_2} = 29 \text{ g/mol}$

$$\frac{V_3}{V_2} = \frac{P_2}{P_3}$$

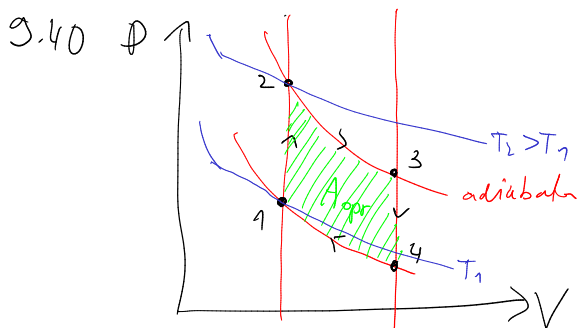
izotermna $\rightarrow pV = p_2 V_2 = p_3 V_3$
 $p = \frac{p_2 V_2}{V}$

$$= \frac{m}{M} R T_2 \ln \frac{P_2}{P_3} = \frac{m}{M} R T_1 \left(\frac{P_1}{P_2} \right)^{\frac{1-\gamma}{\gamma}} \ln \frac{P_2}{P_3}$$

$$Q_{pr} = \frac{1.96 \cdot 8300 \text{ J} \cdot 293 \text{ K} \cancel{\text{ kmol}}}{\cancel{\text{ kmol}} \cdot 29 \text{ kg}} \left(\frac{1}{0.8} \right)^{\frac{1-1.4}{1.4}} \ln \frac{0.8}{1} = \underline{\underline{-18 \text{ kJ}}}$$

$$\Delta W_m = Q_{pr} + A_{pr}$$

$$A_{pr} = \Delta W_m - Q_{pr} = -132 \text{ J} - (-18 \text{ kJ}) = \underline{\underline{5 \text{ kJ}}}$$



$$\kappa = \frac{\gamma}{\gamma - 1} = 1,4 \quad \left| \quad \frac{V_3}{V_2} = 4, 1 = \lambda \quad \begin{matrix} V_1 = V_2 \\ V_3 = V_4 \end{matrix} \right.$$

$\eta = ?$ izohoristički

$$A_{opr} > 0$$

$$A_{opr} = -A_{pr} = \int p dV$$

$$A_{opr}^{1 \rightarrow 2} = 0 \quad dV = 0$$

$$A_{opr}^{2 \rightarrow 3} = \int_{V_2}^{V_3} p dV > 0$$

$$A_{opr}^{3 \rightarrow 4} = 0 \quad dV = 0$$

$$A_{opr}^{4 \rightarrow 1} = \int_{V_4}^{V_1} p dV < 0 \quad \left| \quad A_{opr}^{4 \rightarrow 1} \right| = \int_{V_1}^{V_4} p dV < A_{opr}^{2 \rightarrow 3}$$

$$A_{opr} = A_{opr}^{2 \rightarrow 3} + A_{opr}^{4 \rightarrow 1} > 0$$

$$\Delta W_i = A + Q = -A_{opr} + Q_{dovr} + Q_{odv}$$

$$\eta = \frac{A_{opr}}{Q_{dovr}}$$

$$Q^{1 \rightarrow 2} = \Delta W_m^{1 \rightarrow 2} = m c_V (T_2 - T_1) > 0 \quad \text{dovodena toplota}$$

$$Q^{2 \rightarrow 3} = 0 \quad (\text{adiabata})$$

$$Q^{3 \rightarrow 4} = \Delta W_m^{3 \rightarrow 4} = m c_V (T_4 - T_1) < 0 \quad \text{odvedena toplota}$$

$$Q^{4 \rightarrow 1} = 0 \quad (\text{adiabata})$$

$$A_{opr}^{2 \rightarrow 3} = \int_{V_2}^{V_3} p dV = -m c_V (T_3 - T_2)$$

$$\Delta W_m^{1 \rightarrow 3} = A^{2 \rightarrow 3} = -A_{opr} \quad (Q = 0, \text{ jer je sprazna adiabata})$$

$$A_{opr}^{4 \rightarrow 1} = -m c_V (T_1 - T_4)$$

$$\eta = \frac{A_{opr}^{2 \rightarrow 3} + A_{opr}^{4 \rightarrow 1}}{Q^{1 \rightarrow 2}} = \frac{-m c_V (T_3 - T_2) - m c_V (T_1 - T_4)}{m c_V (T_2 - T_1)} = \frac{T_2 - T_1 + T_4 - T_3}{T_2 - T_1} =$$

$$= 1 - \frac{T_3 - T_4}{T_2 - T_1}$$

2 → 3 adiabata

4 → 1 adiabata

$$\begin{cases} p_2 V_2^\kappa = p_3 V_3^\kappa \\ \therefore \left(\frac{p_2 V_2}{T_2} = \frac{p_3 V_3}{T_3} \right) \end{cases}$$

$$V_2^{\kappa-1} T_2 = V_3^{\kappa-1} T_3$$

$$T_3 = T_2 \left(\frac{V_2}{V_3} \right)^{\kappa-1}$$

$$T_3 = T_2 \left(\frac{1}{\lambda} \right)^{\kappa-1}$$

$$T_2 = T_3 \lambda^{\kappa-1}$$

$$T_1 = T_4 \left(\frac{V_4}{V_1} \right)^{\kappa-1}$$

$$T_1 = T_4 \lambda^{\kappa-1}$$

$$\eta = 1 - \frac{T_3 - T_4}{T_3 \lambda^{\kappa-1} - T_4 \lambda^{\kappa-1}} =$$

$$= 1 - \frac{T_3 - T_4}{\lambda^{\kappa-1} (T_3 - T_4)} =$$

$$= 1 - \lambda^{1-\kappa} = 1 - (4,1)^{-0,4} =$$

$$= \underline{\underline{0,43 = 43\%}}$$